

Study of the compatibility of historical stone materials of the Kuldīga brick-vault bridge across the Venta river and materials used in the restoration of the bridge

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The Kuldīga bridge (1873–1874) is one of the longest brick-vault bridges in Europe (164 m). In the first world war (1915), two parts of the bridge were destroyed by the Russian army. In its previous restoration in the 1930, incompatible materials – dense portland cement based mortars and concrete – were used. In the frame of the recent restoration project (2007–2008), a scientific investigation of the old materials, corrosion products and new compatible materials developed for the restoration was carried out. Based on the obtained results, recommendations and suggestions on materials and methods to be used in the restoration were given to provide for the compatibility of materials and the sustainability of the bridge.

Introduction

The Kuldīga brick-vault bridge across the Venta river is an inseparable element of the old town, one of Kuldīga's symbols, a significant architectural monument. It is the longest (164 m) brick-vault bridge not only in Latvia, but also in Europe. The bridge was built in 1873–1874. It consisted of seven brick-vault span lengths; two of them were destroyed in the First World War (1915) by the Russian army. In 1926, instead of them, were two concrete vault span lengths erected. In the 1930, in the restoration of the bridge incompatible materials were used (concrete and as a joint material dense portland cement based mortars). In 1958, the historical stones of the pavement were covered with bituminous concrete.

To retain the uniqueness and authenticity of the bridge, in 2007–2008 the Kuldīga City Council launched a huge restoration project with the financial support from the European Regional Development Foundation.

An active part in the restoration project of the bridge was taken by scientists of the Centre for Conservation and Restoration of Stone Materials of the Riga Technical University. Stone materials and corrosion products were investigated, new compositions of materials for restoration were developed [1].

After the restoration of the bridge performed by the Kuldīga City Council, this significant success was submitted for the Europe Nostra award.

The aim of this study was to ascertain the type and properties of building materials (historical) of the Kuldīga brick-vault bridge to choose compatible materials to be used in the restoration of the bridge to provide for its sustainability.

Experimental

The methods applied were:

1. Full chemical analysis [2, 3]. The reagents used for all analyses were of “pure for analyses” (p.a.).

purity grade. The chemical composition of the historical materials was detected by wet chemical analysis determining the ignition loss of samples at 1000 °C, followed by the dissolution of a sample in HCl. The obtained solution was further used for soluble SiO₂ and Al₂O₃ determination by sedimentation.

The next step was dissolving the samples in H₂SO₄ and HF for CaO, MgO, Fe₂O₃, Na₂O and K₂O detection:

- CaO and MgO – by complexometric titration with EDTA, using methyl thymol blue and calcein as indicators,
 - Fe₂O₃ by *ФЭК-56ИМ* photocolourimetry using sulphosalicylic acid forming a yellow complex with Fe₂O₃,
 - Na₂O and K₂O by flame photometry *Jenway PFP7*. Butane or propane flame was used to excite the alkalis to emit their characteristic spectrum in the visible range: the emission is proportional to alkali content at low concentrations.
2. XRD – by using X-ray *Rigaku Ultima+* diffractometer, Japan, with Cu K α , radiation at a scanning interval 5–60 °(2 θ) and speed 2 °/min.
 3. Hydro tests (hydraulic properties, water absorption, porosity) [4].
 4. Grain size distribution test [3].

Analysis of the sieve curve was done by dissolving the binder with 14% HCl and sieving the washed aggregate through gradual of finer sieves.

Results and discussion

The general visual observation of the Kuldīga bridge built of seven vault span length has revealed a great difference among its parts: five original brick vaults are in a good state (bricks without damage), while the two restored (in 1934) vaults are in an unsatisfactory state.

The bricks are split, cracked, some dolomite plates are damaged or absent. The mortar of this part, used as joint material, also visually differs from that used in the original part of the bridge. In some places biological damage was seen.

Results of chemical analyses, summarized in Table 1, show that historical mortar used for cladding dolomite

stone and brick vaults was a strongly hydraulic dolomitic lime mortar with the 1 : 2.7 binder-to-filler ratio (sample 6), but in the renovated part cement mortar had been used for joining stones and bricks (samples 3, 4). For the parapet cornices, dolomite plates had been used (sample 8).

Table 1. Chemical composition of historical stone materials, % by weight

Component	Sample				
	3 (concrete)	4 (surface of concrete)	6 (old mortar)	8 (stone)	Error ± absolute %
Ignition loss: at 400 °C at 1000 °C	9.11	8.08	7.90	3.32	0.5
	13.10	13.24	12.54	38.50	0.5
Sand and gravel	41.41	41.90	60.50	4.62	0.5
SiO ₂ soluble	5.04	6.48	2.11	3.80	0.1
CaO	25.21	25.37	11.88	28.30	0.3
MgO	1.93	1.28	5.82	16.19	0.3
Al ₂ O ₃	1.06	1.04	0.96	3.22	0.3
Fe ₂ O ₃	1.88	1.10	1.15	0.88	0.1
Na ₂ O	0.45	0.96	0.46	0.47	0.1
K ₂ O	0.81	1.12	0.78	0.64	0.1
Summary	100.14	99.95	100.56	100.34	–
CaO/MgO	13.1	19.8	2.04	1.74	–
Binder to filler ratio	1 : 1.2	1 : 1.2	1 : 2.7	–	–
Hydraulic modulus $\frac{\text{CaO} + \text{MgO}}{\text{SiO}_2 + \text{Al}_2\text{O}_3 + \text{Fe}_2\text{O}_3}$	6.8	3.09	4.19	–	–
	Cement mortar	Cement mortar	Strongly hydraulic dolomitic mortar	Dolomite stone	

Table 2 lists the contamination of brick, concrete, mortar and dolomite stone by soluble salts. It is obvious that (cleaning of the bridge in 2001) the content of soluble salts is very low and meets the requirements of uncontaminated masonry. The pH analysis of two concrete samples (2, 3) shows that the carbonation of concrete has taken place, resulting in the pH reduced

from 11 to 8.5. In some places, on the surface white deposits were seen. It is possible that CaCO₃ formed due to water migration into concrete, washing out Ca(OH)₂ with the following carbonation. Partially it was cleaned up by sandblast, but this operation is not recommended because it destroys the natural protective layer and causes the further corrosion of material.

Table 2. Contamination of building materials with soluble salts, mass% and pH

Sample	Component				
	Ca ²⁺	Mg ²⁺	Cl ⁻	SO ₄ ²⁻	pH
1. Brick	0.23	–	0.15	–	–
2. Brick	0.47	0.03	0.12	0.49	–
3. Concrete	0.12	0.02	0.03	0.39	9.30
4. Concrete	0.30	–	0.01	–	8.50
5. Brick ~1880	0.07	1.49	0.05	–	–
6. Old mortar	0.08	0.49	0.17	–	8.25
7. Brick ~1880	0.07	1.49	0.17	0.18	–
8. Dolomite	0.26	0.10	0.07	–	–
Error ± absolute %	0.10	0.10	0.05	0.10	0.20

Table 3 presents the results of grain size distribution analysis of concrete (3, 4) and mortar (6). We see that the concrete contains above 0.7 mm about 47.8–57.0% of coarse particles, versus ~72.4% in the historical dolomite lime mortar. Dolomite mortar ensured the porosity of joints and a perfect moisture exchange in masonry with

the environment, i.e. the “breathing” of masonry. It explains the preservation of this part of the bridge to our days. On the contrary, the part of the bridge restored in 1926–1930, when dense portland cement mortars and concrete boarders instead of dolomite stone boarders had been used (samples 3, 4), was in a bad state.

Table 3. Grain size distribution analysis

Grain size D, mm	Sample			
	3 (concrete)	4 (surface of concrete)	6 (old mortar)	Error ± absolute %
> 0.7	57.0	47.8	72.4	0.5
0.5–0.7	22.5	25.0	15.3	0.5
0.2–0.5	9.0	14.4	5.6	0.5
0.16–0.20	2.5	4.2	1.4	0.5
< 0.16	9.0	8.6	5.3	0.5
Summary	100.0	100.0	100.0	–

The degradation of historical stone and brick buildings is known to depend to a great extent on their physical and mechanical properties. It is closely linked to the moisture-related behaviour of the material. The behaviour of the material in response to moisture is generally determined by its microstructure, especially porosity [5].

The water absorption and porosity of historical and new by developed materials used in the restoration of the

bridge are plotted in Figures 1–3. A comparison of these data shows that for building the original Kuldiga bridge, bricks with a high porosity (25–31%) and accordingly water absorption (12–16%) and dolomite with porosity 22% and water absorption 9% were used. As a joining material, compatible hydraulic dolomite lime mortar was used. Materials for the restoration were developed and selected with regard to these characteristics (Figs. 2, 3).

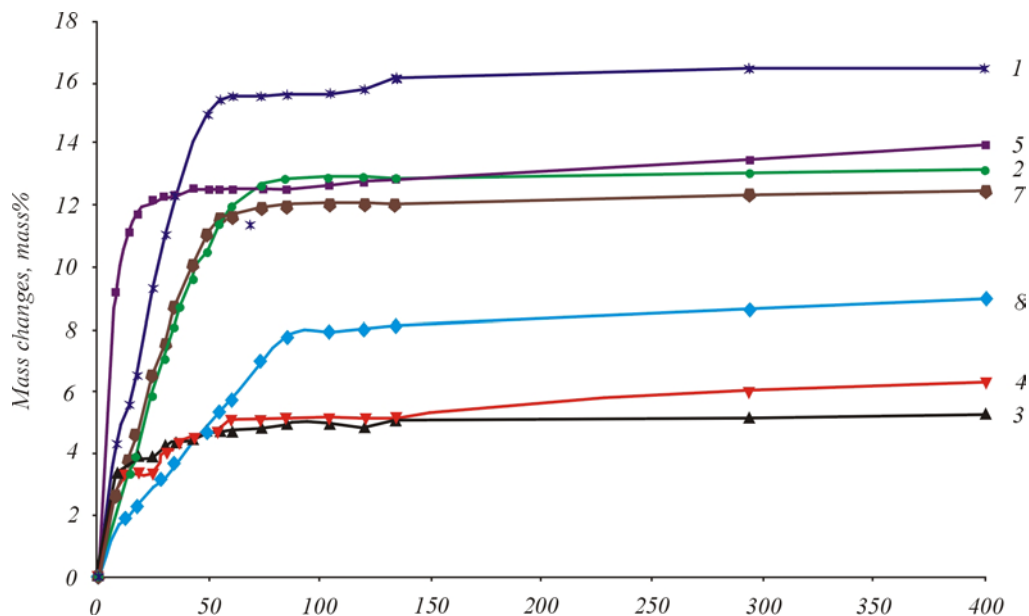


Fig. 1. Water absorption of historical stone materials: 1 – brick; 1926-1930, 2 – brick; 1926-1930, 3 – concrete; 1926-1930, 4 – concrete 1926-1930, 5 – brick; 1874, 7 – brick; 1874, 8 – dolomite; 1874

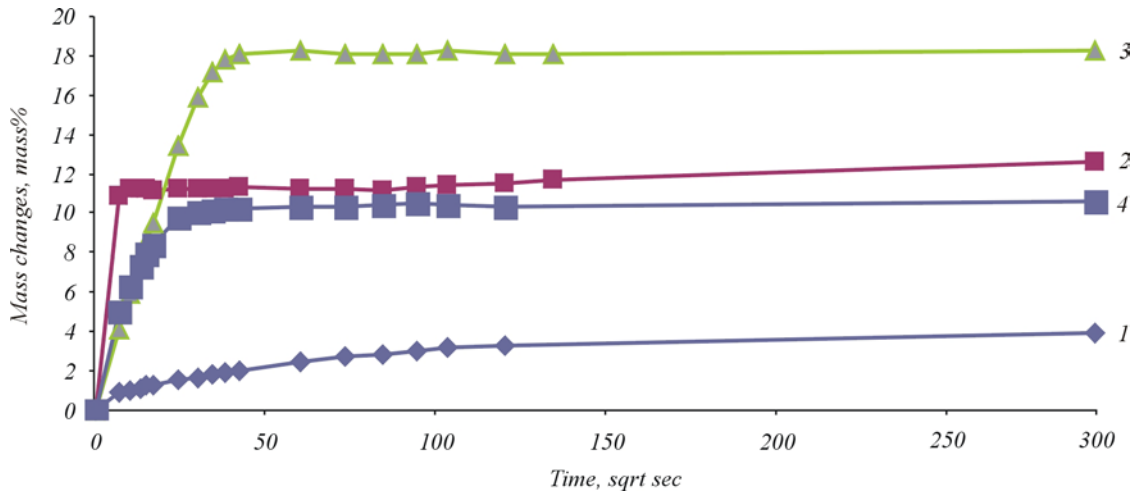


Fig. 2. Water absorption of stone materials used in the restoration: 1 – Estonian dolomite 2008, 2 – brick „Sencis” , 3 – calstic mortar 2008, 7 – dolomitic mortar 2008

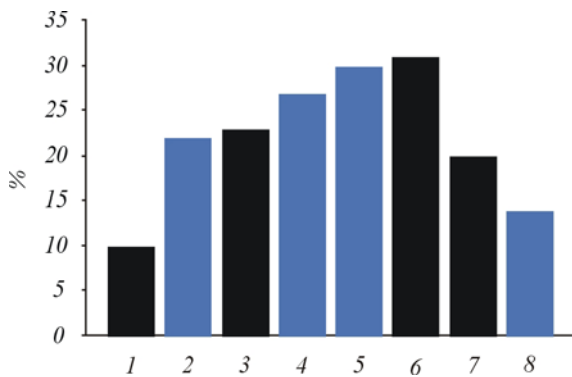


Fig. 3. Porosity of historical and new materials: 1 – Estonian dolomite 2008, 2 – dolomite 1874, 3 – brick „Sencis” 2008, 4 – brick 1880, 5 – brick 1926-1930, 6 – calstic mortar 2008, 7 – dolomitic mortar 2008, 8 – concrete 1926-1930

Figure 4 presents bridge historical mortar XRD used for joints (1) and the newly developed calcium lime mortar for restoration of the Kuldiga bridge (2). It is evident that SiO_2 (quartz), CaCO_3 (calcite), $2\text{CaO}\cdot\text{SiO}_2\cdot 2\text{H}_2\text{O}$ and $\text{CaO}\cdot\text{SiO}_2\cdot\text{H}_2\text{O}$ are present in both samples. The differing phases of samples are MgCO_3 (magnesite – sample 1) and $\text{Ca}(\text{OH})_2$ (portlandite – sample 2). Chemical analysis (Table 1) and XRD pattern 1 confirm that the historical mortar is based on dolomite lime. This explains the preservation of the old brick vaults (1873–1874) to our days, unlike the brick vaults restored in 1926–1930. The presence of $\text{Ca}(\text{OH})_2$ in sample 2 indicated that the carbonization process while forming CaCO_3 had not been complete.

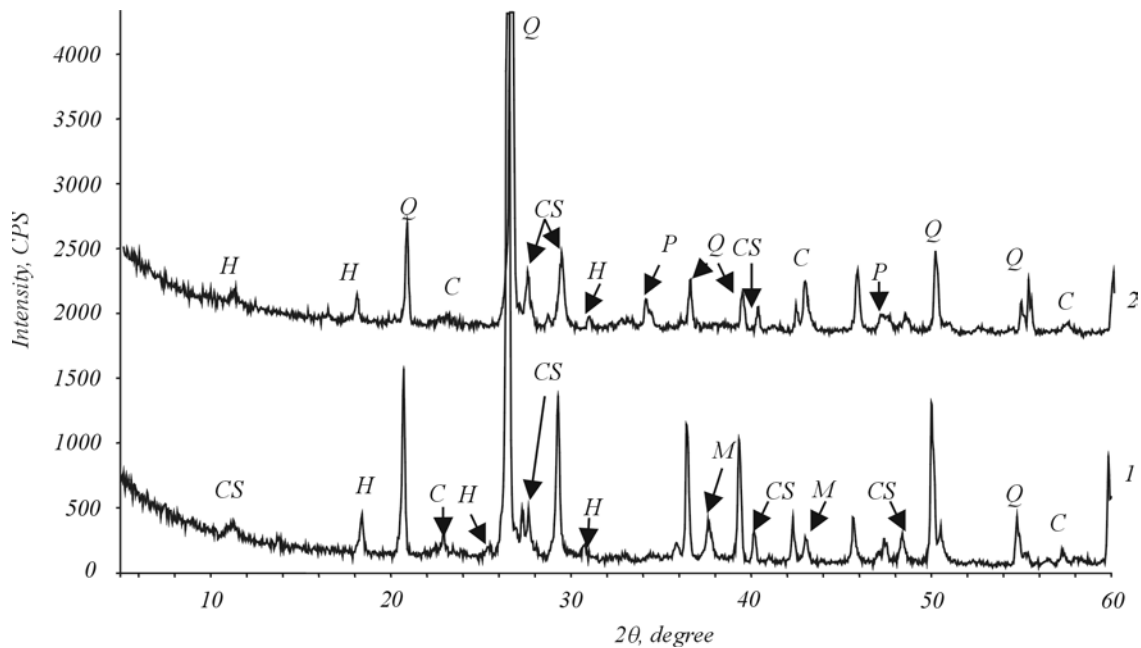


Fig. 4. XRD patterns of historical (1) and newly developed calcium lime mortar (2). Q – quartz, SiO_2 ; C – calcite, CaCO_3 ; P – portlandite, $\text{Ca}(\text{OH})_2$; M – magnesite, MgCO_3 ; H – hydrosilicate, $2\text{CaO}\cdot\text{SiO}_2\cdot 2\text{H}_2\text{O}$; CS – low- basic hydrosilicate, $\text{CaO}\cdot\text{SiO}_2\cdot\text{H}_2\text{O}$

Conclusions

1. In 1930, the restoration of Kuldiga brick-vault bridge built in 1873–1874 was not successful because of incompatibility of old bricks and mortars used for restoration. The main cause of the incompatibility was the different porosity of bricks (25–31%) and cement mortar (13–14%), resulting in the damage of bricks.
2. In 2007–2008, the restoration of the damaged brick vaults was carried out with replacing those hand-made bricks “Sencis” produced at the SIA “Lode”, which were compatible in colour, dimensions and properties with the old bricks.
3. For joining bricks, compositions of calcium and dolomite lime mortars were developed at the laboratory of the Centre for Conservation and Restoration of Stone Materials of the Riga Technical University.
4. The similar porosity and water absorption of both bricks and mortars provides for their compatibility and the sustainability of the renovated bridge.

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ISTORINIO KULDĪGOS SKLIAUTINIO TILTO PER VENTOS UPEĒ MŪRO IR NAUDOTŲ JO RESTAURAVIMUI MEDŽIAGŲ SUDERINAMUMO TYRIMAS

S a n t r a u k a

Kuldīgos skliautinis tiltas (1873–1874 m.) yra vienas iš ilgiausių mūrinių tiltų Europoje (164 m). Pirmojo pasaulinio karo metais (1915) du jo skliautai buvo susprogdinti besitraukiančios Rusijos armijos. Restauruojant 1930 m. šį tiltą buvo naudojamos netinkamos tam medžiagos – portlandcemenčio skiediniai ir betonai. Pagal 2007–2008 m. Restauracijos projektą buvo moksliskai ištirta senų medžiagų, korozijos produktų ir restauracijai naudotinių naujų medžiagų suderinamumas. Remiantis gautais rezultatais pasiūlytos rekomendacijos ir restauracijos metodai užtikrinantys medžiagų suderinamumą bei tilto tvirtumą.